

Seat No.: \_\_\_\_\_

Enrolment No. \_\_\_\_\_

## GUJARAT TECHNOLOGICAL UNIVERSITY

**B.PHARM – SEMESTER – VII • EXAMINATION – WINTER – 2015**

**Subject Code: 2270012**

**Date: 16/12/2015**

**Subject Name: Green Chemistry**

**Time: 10.30 AM to 1.30 PM**

**Total Marks: 80**

### Instructions:

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

- Q.1** (a) Define Green Chemistry with goals and limitations. **06**  
(b) Write down the twelve principles of Green Chemistry. Explain any two with the help of suitable examples. **05**  
(c) Write a note on inception of Green Chemistry. **05**
- Q.2** (a) Write short notes on prevention on waste/byproducts and Atom economy. **06**  
(b) Explain the role of Green Chemistry in polymer industry. **05**  
(c) Define the terms : Green Solvents and Saponification. **05**
- Q.3** (a) Explain the role of Green Chemistry in pharmaceutical industry. **06**  
(b) Write a note on MAOS. **05**  
(c) Catalytic reagents are superior to stoichiometric reagents. Justify with suitable examples. **05**
- Q.4** (a) Give the Green Synthesis of any two : **06**  
i) Paracetamol ii) Catechol iii) Benzyl Broamide  
(b) Explain the Hofmann elimination in water reaction when induced by microwaves. **05**  
(c) Write a note on microwave assisted Diels Alder reaction. **05**
- Q.5** (a) Define ionic liquids and discuss the advantages of water as solvents over the organic solvents. **06**  
(b) Write a note on microwave assisted solid state reactions with any one suitable example. **05**  
(c) Explain the term ‘Solvent less processes’. **05**
- Q. 6** (a) Discuss the future aspects of Green Chemistry in various areas. **06**  
(b) Write a note on Combinatorial Green Chemistry. **05**  
(c) Explain the terms Hazardous solvents. **05**
- Q.7** (a) Discuss any two reactions under sonication taking a suitable example **06**  
i) Strecker synthesis ii) Cannizaro reaction iii) Oxidation reactions  
(b) Describe the role of ‘Clayan’ as a nonmetallic oxidative reagents in various reactions. **05**  
(c) Write a note on Fries rearrangement microwave assisted reaction in water. **05**

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